

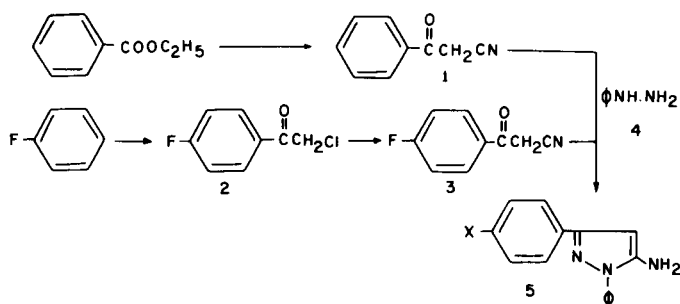
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Nine new fluorine containing 5-amino-1,3-disubstituted pyrazoles have been synthesized and characterized by spectral studies. Condensation reactions of these 5-amino-1,3-disubstituted pyrazoles with fluorinated 1,3-diketones in the presence of glacial acetic acid have been studied and the structures of the resulting 1*H*-pyrazolo[3,4-*b*]pyridines have been confirmed by ir, pmr and ¹⁹F nmr spectral studies.

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The pyrazolopyridine nucleus has aroused great interest in recent years due to a wide variety of biological activities associated with it. Antiinflammatory (1), antipyretic, analgesic (2), bactericidal (3), vasodilating, respiration stimulating, bronchodilating and hypotensive activities (4) of this system have been reported in the literature. The first synthesis of 1*H*-pyrazolo[3,4-*b*]pyridine nucleus was reported by Bülow (5). In our comprehensive program of developing new fluorinated heterocycles, we now report the synthesis and characterization of nine new fluorine-containing 5-amino-1,3-disubstituted pyrazoles and nineteen new fluorine containing 1*H*-pyrazolo[3,4-*b*]pyridines. The preparation of 5-amino-1,3-disubstituted pyrazoles is summarized in the following Scheme.



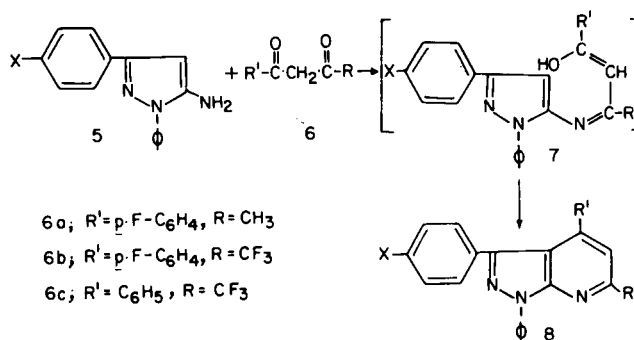
4a; $\phi = p\text{-Cl-C}_6\text{H}_4$
4b; $\phi = p\text{-Br-C}_6\text{H}_4$
4c; $\phi = p\text{-F-C}_6\text{H}_4$
4d; $\phi = \text{C}_6\text{F}_5$
4e; $\phi = p\text{-F-C}_6\text{H}_4\text{SO}_2$
4f; $\phi = p\text{-CH}_3\text{-C}_6\text{H}_4\text{SO}_2$

5a; X=H, $\phi = p\text{-Cl-C}_6\text{H}_4$
5b; X=H, $\phi = p\text{-Br-C}_6\text{H}_4$
5c; X=H, $\phi = p\text{-F-C}_6\text{H}_4$
5d; X=F, $\phi = \text{H}$
5e; X=F, $\phi = \text{C}_6\text{H}_5$
5f; X=F, $\phi = \text{C}_6\text{F}_5$
5g; X=F, $\phi = p\text{-Cl-C}_6\text{H}_4$
5h; X=F, $\phi = p\text{-Br-C}_6\text{H}_4$
5i; X=F, $\phi = p\text{-F-C}_6\text{H}_4$
5j; X=F, $\phi = p\text{-CH}_3\text{-C}_6\text{H}_4\text{SO}_2$
5k; X=F, $\phi = p\text{-F-C}_6\text{H}_4\text{SO}_2$
5l; X=H, $\phi = \text{H}$
5m; X=H, $\phi = \text{C}_6\text{H}_5$

Benzoylacetonitrile **1** was prepared by the condensation of ethyl benzoate with acetonitrile in presence of sodium ethoxide (6,7). 5-Fluorophenacyl chloride **2** was obtained by Friedel-Crafts reaction (8) of fluorobenzene with chloroacetyl chloride and the former gave 5-fluorobenzoylacetonitrile **3** by the nucleophilic replacement of chloride with

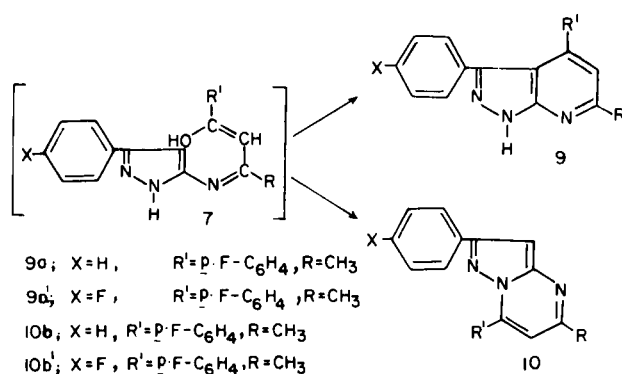
sodium cyanide (9). Formation of compound **3** was confirmed by the appearance of a nitrile absorption peak at 2250 cm^{-1} in the ir spectra. Reactions of **1** and **3** with hydrazine or *p*-substituted arylhydrazines in absolute ethanol (10) gave the corresponding 5-amino-1,3-disubstituted pyrazoles (Table I). The pmr (deuteriochloroform) spectra of **5** exhibit a characteristic methine proton (=CH) resonance signal at δ 6.2, an amino proton resonance signal at δ 3.4 and aromatic proton signals at δ 7.0 to 7.9 ppm. Disappearance of the nitrile absorption peak in the ir spectra from the region of 2250 cm^{-1} and the appearance of a new broad band at $3440\text{-}3210\text{ cm}^{-1}$, due to an amino group, provide firm support for the formation of compounds **5**.

Condensation of the appropriate 5-amino-1,3-disubstituted pyrazole with fluorinated 1,3-diketones (10) in presence of glacial acetic acid yielded the intermediates **7**, which were rapidly cyclised to give 1*H*-pyrazolo[3,4-*b*]pyridines (**8**). Structures of these 1*H*-pyrazolo[3,4-*b*]pyridines have been confirmed by ir, pmr and ¹⁹F nmr spectral studies.



It is interesting to note that the reaction of 5-amino-3-phenylpyrazole or 5-amino-3-(4'-fluorophenyl)pyrazole with 1-(4'-fluorophenyl)-4,4,4-trifluorobutane-1,3-dione yields only one product. However, with 1-(4'-fluorophenyl)butane-1,3-dione, two isomeric products, **9** and **10**, were obtained. These two isomers were separated by column chromatography on silica-gel, using different ratios of benzene ethyl acetate. Structures of these two isomers

have been confirmed with the help of pmr and ir spectral studies. Isomer **9** is yellow in colour and can be formed if cyclization of the intermediate **7** takes place at position-4. The ir spectra of these compounds exhibit a characteristic =NH stretching mode at 3100 cm^{-1} . The pmr spectra also show a =NH resonance signal at δ 8.0 ppm, thus confirming their pyrazolo-pyridine nature. Isomer **10** is white in colour and can be formed on cyclization of the intermediate **7** at position-1. These isomers **10** are devoid of the =NH absorption peak in the ir spectra in the region of 3100 cm^{-1} . The pmr spectrum of **10b** (trifluoroacetic acid) shows a similar a_2, b_2 pattern for both fluorophenyl groups, suggesting a pyrazolo[1,5-a]pyrimidine structure.



EXPERIMENTAL

Melting points are uncorrected. Infrared spectra were recorded using a Perkin Elmer Model 337 spectrophotometer. Proton magnetic resonance spectra and fluorine magnetic resonance spectra were recorded on a Perkin Elmer Model RB-12 spectrometer using tetramethylsilane (TMS) and trifluoroacetic acid (TFA) as internal and external standards. The chemical shifts are reported in ten parts per million and hundred parts per million, respectively.

Benzoylacetone nitrile (**1**).

Benzoylacetone nitrile **1** was prepared by treating ethyl benzoate (150 g., 1 mole) with methyl cyanide (51 g., 1.25 mole) in the presence of sodium ethoxide (68 g., 1 mole). It was isolated in the usual manner (6,7) and was purified by recrystallization from ethanol, m.p. 80° .

4-Fluorophenacyl Chloride (**2**).

Compound **2** was prepared by the chloroacetylation of fluorobenzene (16 ml., 0.1 mole) with chloroacetyl chloride (19.8 ml., 0.1 mole) in the presence of anhydrous aluminium chloride (50 g.) in dry carbon disulfide. 4-Fluorophenacyl chloride was isolated and purified as mentioned earlier. It was crystallized from ethanol, m.p. 45° .

4-Fluorobenzoylacetone nitrile (**3**).

To an ethanolic solution of 4-fluorophenacyl chloride (1.79 g., 0.01 mole), an aqueous solution of sodium cyanide (1.47 g., 0.03 mole) was added dropwise. The reaction mixture was kept as such for four hours, diluted with water (50 ml.) and again kept for an additional four hours. The solution was filtered and the filtrate made acidic with acetic acid. It was cooled and filtered, m.p. 80° , yield 56%.

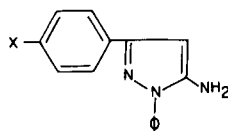
Anal. Calcd. for C_9H_8FNO : N, 8.58. Found: N, 8.42.

4-Fluorophenylhydrazine (**4c**).

4-Fluoroaniline (10 g., 0.1 mole) was diazotized with sodium nitrite in the usual manner (11) and the diazonium solution was filtered. This was added to a solution of sodium sulfite (36 g. in 70 ml. of water) and

Table I

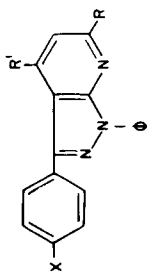
5-Amino-1,3-Disubstituted Pyrazoles



Compound No.	X	Φ	Yield %	M.p. $^\circ\text{C}$	Recrystallization Solvent	Formula	Analysis %	
							Calcd. N	Found N
5a	H	<i>p</i> -Cl-C ₆ H ₄	80	148	Ethanol	C ₁₅ H ₁₂ ClN ₃	15.58	15.32
5b	H	<i>p</i> -Br-C ₆ H ₄	90	130	Ethanol	C ₁₅ H ₁₂ BrN ₃	13.37	13.01
5c	H	<i>p</i> -F-C ₆ H ₄	85	133	Methanol	C ₁₅ H ₁₂ FN ₃	16.66	16.50
5d	F	H	80	130	Ethanol	C ₉ H ₈ FN ₃	23.73	23.60
5e	F	C ₆ H ₅	75	148	Ethanol	C ₁₅ H ₁₂ FN ₃	16.60	16.49
5f	F	C ₆ F ₅	85	135	Methanol	C ₁₅ H ₇ F ₆ N ₃	12.24	12.01
5g	F	<i>p</i> -Cl-C ₆ H ₄	75	135	Ethanol	C ₁₅ H ₁₁ ClFN ₃	14.60	14.50
5h	F	<i>p</i> -Br-C ₆ H ₄	78	139	Methanol	C ₁₅ H ₁₁ BrFN ₃	12.65	12.60
5i	F	<i>p</i> -F-C ₆ H ₄	80	145	Methanol	C ₁₅ H ₁₁ F ₂ N ₃	15.50	15.00
5j	F	<i>p</i> -CH ₃ -C ₆ H ₄ SO ₂	88	190	Ethanol	C ₁₆ H ₁₄ FN ₃ SO ₂	12.68	12.51
5k	F	<i>p</i> -F-C ₆ H ₄ SO ₂	78	170	Methanol	C ₁₅ H ₁₁ F ₂ N ₃ SO ₂	12.52	12.21
5l (a)	H	H	60	124	Ethanol	C ₉ H ₈ N ₃	—	—
5m (a)	H	C ₆ H ₅	58	123	Methanol	C ₁₅ H ₁₃ N ₃	—	—

(a) These compounds have been reported previously by Grandberg, *et al.* (17)

Table II
Analytical Data of 1*H*-Pyrazolo[3,4-*b*]pyridines



Compound No. (a)	x	phi	R'	R	Yield %	M.p. °C	Formula	Calcd			Analysis			Found		
								C	H	F	N	C	H	F	N	C
1a (b)	C ₆ H ₅	H	<i>p</i> -F-C ₆ H ₄	CH ₃	70	143	C ₁₉ H ₁₄ FN ₃	75.25	4.62	6.27	13.86	74.98	4.43	6.01	13.45	
1b	C ₆ H ₅	H	<i>p</i> -F-C ₆ H ₄	CH ₃	25	171	C ₁₉ H ₁₄ FN ₃	75.25	4.62	6.27	13.86	74.98	4.43	6.01	13.45	
2	C ₆ H ₅	H	<i>p</i> -F-C ₆ H ₄	CF ₃	80	194	C ₁₉ H ₁₁ F ₃ N ₃	63.87	3.08	21.29	11.76	63.70	3.00	21.10	11.72	
3	C ₆ H ₅	<i>p</i> -Cl-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CH ₃	73	151	C ₂₅ H ₁₇ ClFN ₃	72.56	4.11	4.59	10.16	72.45	4.10	4.40	10.01	
4	C ₆ H ₅	<i>p</i> -Cl-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CF ₃	78	170	C ₂₅ H ₁₄ ClF ₃ N ₃	64.17	2.99	16.26	8.99	64.00	2.69	16.02	8.49	
5	C ₆ H ₅	<i>p</i> -Cl-C ₆ H ₄	CF ₃	CF ₃	85	130	C ₂₀ H ₁₀ ClF ₆ N ₃	54.36	2.26	25.82	9.51	54.10	2.16	25.43	9.60	
6	C ₆ H ₅	<i>p</i> -Br-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CH ₃	80	140	C ₂₅ H ₁₇ BrFN ₃	65.50	3.71	4.15	9.17	65.31	3.42	4.01	9.00	
7	C ₆ H ₅	<i>p</i> -Br-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CF ₃	65	176	C ₂₅ H ₁₄ BrF ₃ N ₃	58.58	2.77	14.83	8.20	58.28	2.53	14.30	8.01	
8	C ₆ H ₅	<i>p</i> -Br-C ₆ H ₄	CF ₃	CF ₃	80	162	C ₂₀ H ₁₀ BrF ₆ N ₃	49.38	2.06	23.45	8.85	49.31	2.10	23.21	8.45	
9a (b)	<i>p</i> -F-C ₆ H ₄	H	<i>p</i> -F-C ₆ H ₄	CH ₃	65	180	C ₁₉ H ₁₃ F ₂ N ₃	71.02	4.04	11.83	13.08	71.04	4.08	11.45	12.98	
9b	<i>p</i> -F-C ₆ H ₄	H	<i>p</i> -F-C ₆ H ₄	CH ₃	30	210	C ₁₉ H ₁₃ F ₂ N ₃	71.02	4.04	11.83	13.08	71.04	4.08	11.45	12.98	
10	<i>p</i> -F-C ₆ H ₄	H	<i>p</i> -F-C ₆ H ₄	CF ₃	50	183	C ₁₉ H ₁₀ F ₃ N ₃	60.80	2.67	25.33	11.20	60.44	2.68	25.13	11.02	
11	<i>p</i> -F-C ₆ H ₄	H	CF ₃	CF ₃	75	158	C ₁₄ H ₈ F ₇ N ₃	48.13	1.72	38.11	12.03	48.00	1.53	38.01	11.99	
12	<i>p</i> -F-C ₆ H ₄	<i>p</i> -Cl-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CH ₃	80	165	C ₂₅ H ₁₆ ClF ₂ N ₃	69.52	3.70	8.80	9.73	69.42	3.51	8.70	9.61	
13	<i>p</i> -F-C ₆ H ₄	<i>p</i> -Cl-C ₆ H ₄	CF ₃	CF ₃	83	166	C ₂₀ H ₉ ClF ₆ N ₃	52.23	1.96	28.94	9.14	52.25	1.81	28.54	9.01	
14	<i>p</i> -F-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄ SO ₂	<i>p</i> -F-C ₆ H ₄	CF ₃	65	170	C ₂₅ H ₁₃ F ₃ N ₃ SO ₂	56.29	2.44	21.39	7.88	56.19	2.31	21.31	7.77	
15	<i>p</i> -F-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄ SO ₂	CF ₃	CF ₃	68	158	C ₂₀ H ₈ F ₃ N ₃ SO ₂	47.35	1.77	29.98	8.28	47.15	1.70	29.81	8.18	
16	<i>p</i> -F-C ₆ H ₄	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂	<i>p</i> -F-C ₆ H ₄	CH ₃	78	160	C ₂₆ H ₁₉ F ₂ N ₃ SO ₂	65.69	4.00	8.00	8.83	65.58	3.99	8.02	8.71	
17	<i>p</i> -F-C ₆ H ₄	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂	<i>p</i> -F-C ₆ H ₄	CF ₃	70	150	C ₂₆ H ₁₆ F ₂ N ₃ SO ₂	58.98	3.03	17.96	7.93	58.88	3.00	17.66	7.83	
18	<i>p</i> -F-C ₆ H ₄	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂	C ₆ H ₅	CF ₃	85	160	C ₂₈ H ₁₇ F ₂ N ₃ SO ₂	61.05	3.32	14.88	8.22	61.00	3.31	14.63	8.01	
19	<i>p</i> -F-C ₆ H ₄	C ₆ F ₅	CF ₃	CF ₃	88	175	C ₂₈ H ₅ F ₁₂ N ₃	46.60	0.97	44.27	8.16	46.51	0.81	44.17	8.00	

(a) a = Pyrazolo[3,4-*b*]pyridine derivatives; b = Pyrazolo[1,5-*d*]pyrimidine derivatives. (b) In these cases, two isomeric products were isolated and characterized.

Table III
Pmr and ¹⁹F Nmr Data for the Pyrazolo[3,4-b]pyridine Derivatives

Compound No.	x	φ	R'	R	NH	Pmr Spectral Data			Chemical Shift, ppm (δ)			¹⁹ F Nmr Spectral Data	
						CH ₃	Aromatic Protons	Solvent	Ar-F	CF ₃	C ₆ F ₅	Solvent	
1a (a)	C ₆ H ₅	H	<i>p</i> -F-C ₆ H ₄	CH ₃	7.9	2.5	6.9 to 7.8	TFA	36.4	—	—	Acetone	
1b	C ₆ H ₅	—	<i>p</i> -F-C ₆ H ₄	CH ₃	—	2.9	7.0 to 8.0	TFA	39.1	—	—	Acetone	
2	C ₆ H ₅	H	<i>p</i> -F-C ₆ H ₄	CF ₃	8.0	—	7.0 to 7.9	TFA	34.4	-6.8	—	DMSO	
3	C ₆ H ₅	<i>p</i> -Cl-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CH ₃	—	2.6	6.5 to 7.5	TFA	33.1	—	—	DMSO	
4	C ₆ H ₅	<i>p</i> -Cl-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CF ₃	—	—	7.0 to 7.9	DMSO	33.8	-7.4	—	DMSO	
5	C ₆ H ₅	<i>p</i> -Cl-C ₆ H ₄	CF ₃	CF ₃	—	—	7.0 to 8.0	DMSO	—	-7.8	—	Acetone	
6	C ₆ H ₅	<i>p</i> -Br-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CH ₃	—	2.5	6.9 to 7.9	TFA	33.8	—	—	Benzene	
7	C ₆ H ₅	<i>p</i> -Br-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CF ₃	—	—	6.5 to 7.6	TFA	35.6	-15.3	—	Acetone	
8	C ₆ H ₅	<i>p</i> -Br-C ₆ H ₄	CF ₃	CF ₃	—	—	7.1 to 8.0	TFA	—	-8.2	—	Acetone	
9a (a)	<i>p</i> -F-C ₆ H ₄	H	<i>p</i> -F-C ₆ H ₄	CH ₃	8.0	2.6	6.8 to 7.8	TFA	32.1	—	—	Acetone	
9b	<i>p</i> -F-C ₆ H ₄	—	<i>p</i> -F-C ₆ H ₄	CH ₃	—	2.8	7.1 to 8.1	TFA	33.4	—	—	Acetone	
10	<i>p</i> -F-C ₆ H ₄	H	<i>p</i> -F-C ₆ H ₄	CF ₃	7.9	—	7.0 to 8.1	DMSO	35.4	-7.4	—	DMSO	
11	<i>p</i> -F-C ₆ H ₄	H	CF ₃	CF ₃	8.2	—	6.9 to 7.8	TFA	38.0	-5.2	—	Acetone	
12	<i>p</i> -F-C ₆ H ₄	<i>p</i> -Cl-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄	CH ₃	—	2.8	7.1 to 7.9	TFA	33.4	-5.9	—	Acetone	
13	<i>p</i> -F-C ₆ H ₄	<i>p</i> -Cl-C ₆ H ₄	CF ₃	CF ₃	—	—	7.2 to 8.1	TFA	33.8	-7.2	—	Benzene	
14	<i>p</i> -F-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄ SO ₂	<i>p</i> -F-C ₆ H ₄	CF ₃	—	—	7.0 to 7.9	TFA	39.2	-14.6	—	Benzene	
15	<i>p</i> -F-C ₆ H ₄	<i>p</i> -F-C ₆ H ₄ SO ₂	CF ₃	CF ₃	—	—	6.8 to 8.0	TFA	33.2	-8.4	—	Benzene	
16	<i>p</i> -F-C ₆ H ₄	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂	<i>p</i> -F-C ₆ H ₄	CH ₃	—	2.6	7.1 to 8.0	TFA	34.2	—	—	Acetone	
17	<i>p</i> -F-C ₆ H ₄	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂	<i>p</i> -F-C ₆ H ₄	CF ₃	—	—	7.2 to 7.9	TFA	38.2	-6.2	—	Acetone	
18	<i>p</i> -F-C ₆ H ₄	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂	C ₆ H ₅	CF ₃	—	—	6.8 to 8.0	TFA	32.6	-6.9	—	Benzene	
19	<i>p</i> -F-C ₆ H ₄	C ₆ F ₅	CF ₃	CF ₃	—	—	6.9 to 8.1	TFA	36.5	—	—	Acetone	
									32.4	—	—	Acetone	
									38.6	-8.2	—	Acetone	
									32.7	—	—	Acetone	
									32.9	—	—	Acetone	
									33.6	-9.2	—	Acetone	
									36.8	-8.4	71	Acetone	
									—	-9.2	79	Acetone	
									—	—	88	Acetone	

(a) In these cases, two isomeric products were isolated and characterized.

potassium hydroxide (4 g.). The resultant bright orange solution was warmed at 60° for one hour and then filtered. The filtrate was acidified with hydrochloric acid (8 ml.) and again heated for six hours. On addition of hydrochloric acid (50 ml.), 4-fluorophenylhydrazine hydrochloride crystallized as tiny needles, m.p. 36°.

Sulphonyl Hydrazide (4e,f).

4-Fluorobenzenesulphonyl chloride or 4-toluenesulphonyl chloride (1 mole) was dissolved in ethanol (15 ml.) and added with stirring to hydrazine hydrate (80%, 2 mole), in ethanol (10 ml.) (12). The solid obtained was isolated in the usual way and recrystallized from hot water. Compound 4e had m.p. 90° (lit. (13) m.p. 90-92°). Compound 4f had m.p. 110° (14).

5-Amino-1,3-disubstituted Pyrazoles (5a-k).

Benzoylacetonitrile or 4-fluorobenzoylacetonitrile (0.92 mole) was dissolved in absolute ethanol (2000 ml.), and the appropriate phenylhydrazine (0.92 mole) was added to it. The reaction mixture was refluxed for six to eight hours and the excess solvent was removed under reduced pressure. The crude product was crystallized from ethanol, yields 70-90%. All of synthesized compounds, along with their analytical data, are given in Table I.

1-(4'-Fluorophenyl)-4,4,4-trifluorobutane-1,3-dione (6b).

These fluorinated-1,3-diketones were prepared by the procedure of Joshi, *et al.* (15), by treating the fluorinated aryl ketone (0.1 mole) with the appropriate ester (0.2 mole) in the presence of sodamide (7.8 g., 0.2 mole) in dry ether. The resultant sodium salt of the 1,3-diketone was dissolved in water and aqueous layer was acidified with hydrochloric acid (10%) yielding the corresponding 1,3-diketones. These 1,3-diketones were recrystallized from petroleum ether. Compound 6a had m.p. 51°. Compound 6b had m.p. 40° (lit. (16) m.p. 40-42°). Compound 6c had m.p. 40°.

1-(4'-Chlorophenyl)-4,6-difluoromethyl-3-(4'-fluorophenyl)pyrazolo[3,4-b]pyridine (13).

5-Amino-1,3-disubstituted pyrazole (0.2 mole) was dissolved in glacial acetic acid (100 ml.) and fluorinated 1,3-diketone (0.2 mole) was added to it. The reaction mixture was heated to reflux for 10 to 12 hours on a sand bath. Excess of solvent was removed under reduced pressure and the compound recrystallized from ethanol. All synthesized compounds along with their analytical data are given in Table II and all spectral data are reported in Table III.

Reaction of 5d and 5l with 6a resulted in a solid which gave two spots on tlc (benzene:ethyl acetate; 50:50). The solids were chromatographed on silica gel, eluting with benzene-ethyl acetate, yielded two isomeric

products (9 and 10). The initial fraction (benzene:ethyl acetate; 75:25) yielded yellow solids which, after recrystallization from methanol, were identified as pyrazolo[3,4-b]pyridines (9a and 9a'). The second fraction (benzene:ethyl acetate; 50:50) yielded white solid by-products, which were recrystallized from ethanol (55%) and were identified as pyrazolo[1,5-a]pyrimidines (10b and 10b').

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